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Question 4 Answer: C Explanation: $3\text{KOH (aq)} + \text{AlCl}_3 \text{ (aq)} \rightarrow 3\text{KCl (aq)} + \text{Al(OH)}_3 \text{ (s)}$
 $3\text{Zn(NO}_3)_2 \text{ (aq)} + 2\text{K}_3\text{PO}_4 \text{ (aq)} \rightarrow \text{Zn}_3\text{(PO}_4)_2 \text{ (s)} + 6\text{KNO}_3 \text{ (aq)}$
 $\text{Sr(OH)}_2 + 2\text{NH}_4\text{CH}_3\text{COO (aq)} \rightarrow 2\text{NH}_4\text{OH} + \text{Sr(CH}_3\text{COO)}_2 \text{ (aq)}$
 $\text{Na}_2\text{SO}_4 \text{ (aq)} + \text{Pb(NO}_3)_2 \text{ (aq)} \rightarrow \text{PbSO}_4 \text{ (s)} + 2\text{NaNO}_3 \text{ (aq)}$ Question 5

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Explanation: Use kinematics equations to find the initial vertical velocity, given $v_{\text{vert}} = 0$ at 1.8 sec after launch:

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$v = u + at$ $0 = u - 10 \times 1.8$ $u = 18 \text{ms}^{-1}$. Then, use trigonometry to determine v at 35° : $v = 18 \cos(55^\circ) = 10.3 \text{ms}^{-1}$. Question 18. Answer: $R = 46 \text{ m}$
Explanation: $v_{\text{horiz}} = v \times \cos(35^\circ) = 10.3 \text{ms}^{-1}$. $x = v_{\text{horiz}} \times t = 46.26 \text{ m}$.

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Units 3 & 4 – Written examination (TSSM 's 2015 trial exam updated for the current study design)

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Section A - Multiple Choice (1 mark each) Question 1

Answer: C Explanation: The density of lines at a specific location in space reveals information about the strength of ...

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2011 Trial Examination SOLUTIONS SECTION A – Core Area of Study 1 – Motion ... 2011 PHYS EXAM 2

TSSM 2011 Page 5 of 21 Question 12 Answer: $6.7 \times 10^4 \text{ N m}^{-1}$

Explanation: The gravitational potential energy at 5 m ($mgh = 0.15 \times 10 \times 5 = 7.5 \text{ J}$) is all converted to kinetic ...

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2010 Trial Examination SOLUTIONS SECTION A – Core Area of Study 1- Electric Power Question 1

Answer: B ... 2010 PHYS EXAM 2 TSSM 2010 Page 3 of 16 Question 10 Answer: 300 turns Explanation:

Current at the generator is $10000/250 = 40 \text{ A}$. Current in the lines is 10 A. Thus the transformer is a step-up in the ratio of 1:4 ...

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time: 1 hour and 30 minutes QUESTION & ANSWER

BOOK Structure of book Section Number of questions

Number of questions to be answered Number of marks

A 20 20 20 B 6 6 66 Total 86 Students are ...

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BOX IS ...

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SECTION A: Multiple-choice questions (1 mark each)

Question 1 Answer: D Explanation: Ice is less dense

than water as the regular array of water molecules

formed as a result of hydrogen bonding causes gaps to

form. Question 2 Answer: C Explanation: $2140 \text{ J} = 2.14$

$\text{kJ mol}^{-1}(\text{water}) = 2.14/6.0 = 0.357 \text{ mol mol}^{-1}(\text{water}) =$

$0.357 \times 18.0 = 6.42 \text{ g}$ Question 3 Answer: D ...

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TSSM 2011 Page 1 of 19 PHYSICS Unit 1 – Written examination 1 2011 Trial Examination SOLUTIONS SECTION A Area of Study 1 – Nuclear Physics and Radioactivity Question 1 Answer:A Explanation: As per definition of each type of radioactive decay. Alpha is the heaviest and doubly

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