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Problem SC2.1. Problem SC2.2. Problem SC3.1. Clockwise. Problem SC3.2. Counter-clockwise. Problem SC3.3. Counter-clockwise. Problem SC3.4. Clockwise. Problem SC3.5. Enantiomer B has a molecular weight of 126 g/mol, a density of 0.995 g/mL, an optical rotation of  $[\alpha] = -26^\circ$ , a melting point of  $65^\circ\text{C}$  and a boiling point of  $225^\circ\text{C}$ . Problem SC3.6.

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Minimize  $90S_1 + 110S_2 + 105S_3 + 200W_1 + 250W_2 + 240W_3 - (60R_1 + 65R_2 + 68R_3 + 150V_1 + 175V_2 + 200V_3) - 7(T_1 + T_2 + X_1 + X_2)$   
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SOLUTIONS OF SELECTED PROBLEMS Problem 36, p. 63 If  $\mu(E_n) < \infty$  and  $\chi_{E_n} \rightarrow f$  in  $L^1$ , then  $f$  is a.e. equal to a characteristic function of a measurable set. Solution: By Corollary 2.32, there exists a subsequence  $\chi_{E_{n_j}} \rightarrow f$  a.e.. Let  $A = \{x : \chi_{E_{n_j}}(x) \rightarrow f(x)\}$ . For  $x \in A$ ,  $\{\chi_{E_{n_j}}(x)\}$  is a sequence of 0's and 1's, so its limit can be either 0 or 1.

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Solutions to Selected Problems. Answer to first problems on page 3. What is the 90% confidence interval for BMI? (Note that  $Z=1.645$  to reflect the 90% confidence level.)

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Problem C01.1. Problem C01.2. Problem C01.3.

a) The double bond means two pairs of electrons are shared between the carbon and oxygen, instead of just one. As a result, the oxygen is able to pull more electron density away from the carbon. The carbon becomes much more positive in this case than in the ...

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9-1. Appendix A  $FV = PV FVIF$  a. \$3,000 1.12

b. \$3,360 1.12 c. \$3,763

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127.83) ===== Answer to BMI Problem on page 3. Question: Using ...

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