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bottom atoms are

in layer A, the

center atom in

layer B. $1/6$ of

each atom at

120° corners is

inside the unit

cell, $1/12$ of

each atom at 60°

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corners is
inside the unit
cell; the atom
in the center of
course is
entirely in the
unit cell.

Adding these up,
we have 2
atoms/unit cell.

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