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A substance combines with oxygen, releasing a large amount of energy as heat and light, in a(n) . 18. The decomposition of a substance by an electric current is called . 19. A(n) orders the elements by the ease with which they undergo certain chemical reactions. 20.

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continued 18. When a strong acid is
titrated with a weak base, the pH of the

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With Answer Key
solution at the equivalence point is than 7.
19. A is a highly purified solid used to
check the concentration of a standard
solution. 20. A 1 M solution of NaOH will
have a pH that is

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Modern Chemistry 137 Chapter Test
Name Class Date Chapter Test A,
continued _____ 19. In the figure on the
previous page, the pH at the equivalence
point a. is equal to 7.0. b. is greater than
7.0. c. is less than 7.0. d. cannot be
determined from the data given. _____ 20.
In the figure on the previous page, the
volume of titration standard

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Modern Chemistry 105 Chapter Test
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continued Use this figure to answer
questions 7 and 8. _____ 7. A solution

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containing 35 g of Li_2SO_4 dissolved in 100 g of water is heated from 10°C to 90°C . According to information in the figure, this temperature change would result in a. an additional 5 g of Li_2SO_4 in ...

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18. Dalton's atomic theory acyreed with the modern atomic theory EXCEPT for the statement that all atoms of the same element have the same Complete the following table to compare the types of subatomic particles. 19. 20. 21. 10 Particle

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The basis of an ionic bond is the__
absorption of water into their solid
structures. Elements tend to gain or lose
electrons in order to acquire the electron
configuration of a__ noble gas. Ions that
are formed from more than one atom are
called__

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Action. Fluid. Surface Tension.

Vaporization. the attraction of the surface
of a liquid to the surface of a... a substance
that can flow and therefore take the shape
of its... a force that tends to pull adjacent
parts of a liquid's surfac....

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