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Chapter 7 test.

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Answers

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basis of an

ionic bond is

the__ absorption

of water into

their solid

structures.

Elements tend to

gain or lose

electrons in

order to acquire

the electron

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configuration of
a__ noble gas.
Ions that are
formed from more
than one atom
are called__

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formula.

indicates the

relative number

of atoms of each

kind in a

chemical

compound.

Reveals the

number of atoms

of each element

contained in a

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ANSWER Answer

the following

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1. Label each of the following statements as True or False:

- True a. If the formula mass of one molecule is x amu, the molar mass is x g/mol.
- False b. Samples of equal numbers of moles of two different chemicals

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7 Chemical

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Name Section

Quiz, continued

Class Date 6.

The Lewis

structure of

which of the

following

compounds does

not follow the

octet rule? a.

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NH₃ b BF₃ c. d.

H₂O 7. In a
double covalent
bond, a. one

atom has more
than eight
valence
electrons.

*Home - David
Brearley High
School*

CHAPTER 5 REVIEW

The Periodic Law

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Chemistry SECTION 1 SHORT

ANSWER Answer

Chapter 7 Quiz
the following

Answers
questions in the

space provided.

1. c In the modern periodic table, elements are ordered (a) according to decreasing atomic mass. (b) according to Mendeleev's

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Answers
Derutारे

original design.

(c) according to
increasing
atomic number.

(d) based on
when they were
discovered. 2. d

Mendeleev
noticed that
certain
similarities in
the ...

5 The Periodic

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137 Chapter Test

Answers
Name Class Date

Chapter Test A,
continued

_____19. In the figure on the previous page, the pH at the equivalence point a. is equal to 7.0. b. is greater than

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7.0. c. is less
than 7.0. d.
cannot be
determined from
the data given.

_____20. In the
figure on the
previous page,
the volume of
titration
standard

Assessment

Chapter Test A -

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