

## Modern Chemistry Chapter 12 Review Answers

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Modern Chemistry Chapter 12 Review Answers Liquids And Solids

Modern Chemistry 1 Solutions CHAPTER 12 REVIEW Solutions Teacher Notes and Answers Chapter 12 SECTION 1 SHORT ANSWER 1. c 2. a 3. b 2. a. alcohol b. water c. the gels 3. The mixture is a colloid. The properties are consistent with those reported in Table 3 on page 404 of the text. The particle size is small, but not too small, and the mixture

CHAPTER 12 REVIEW Solutions - Weebly

CHAPTER 12 REVIEW Solutions MIXED REVIEW SHORT ANSWER Answer the following questions in the space provided. 1. Solid CaCl<sub>2</sub> does not conduct electricity. Explain why it is considered to be an electrolyte. CaCl<sub>2</sub> is an ionic solid. In the crystal form, its ions are locked in position. Dissolving

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Modern Chemistry Chapter 12-Solutions Section 1- Types of Mixtures Solutions are homogeneous mixtures of two or more substances in a single phase. Soluble describes a substance as capable of being dissolved. Solvent is the dissolving medium in a solution. Solute is the substance that is dissolved in a solution.

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