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~~SHORT ANSWER Answer the following questions in the space provided. 1. Indicate whether each of the following is a physical or chemical property of water.~~

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CHAPTER 10 REVIEW States of Matter SECTION 3 SHORT ANSWER Answer the following questions in the space provided Modern chemistry chapter 10 review answers states of matter section 1. 1. Match description on the right to the correct crystal type on the left Modern chemistry chapter 10 review answers states of matter section 1. b ionic crystal (a) has mobile electrons in the crystal c covalent ...

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Modern Chemistry Chapter 10 States of Matter. Modern Chemistry Chapter 10 States of Matter Kinetic-Molecular Theory The kinetic-molecular theory of matter is based on the idea that particles of matter are always in motion and this motion has consequences that affect its physical properties.

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or a real gas ideal gas Modern Chemistry Chapter 10 Review Answers States Of water vapor MODERN CHEMISTRY STATES OF MATTER 87 of matter section 5 The molar enthalpy of vaporization of water is  $\text{kJ mol}^{-1}$  and the molar enthalpy of fusion States of Matter Weebly CHAPTER 10 Modern chemistry chapter 10 review answers states of matter section 5. . .

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Chemistry Chapter 10. diffusion. effusion. Graham's law of effusion. permeability. the spontaneous, uniform spreading and mixing of particles caused by a process in which gas particles pass through a tiny opening in a barrier. the gas law that states that the rate of effusion is inversely proportional to the square root of the molar mass.

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CHAPTER 10 REVIEW States of Matter SECTION 4 SHORT ANSWER Answer the following questions in the space provided. 1. a When a substance in a closed system undergoes a phase change and the system reaches equilibrium, (a) the two opposing changes occur at equal rates. (b) there are no more phase changes. (c) one phase change predominates.

10 States of Matter - Ms. Agostine's Chemistry Page

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