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Honors Chemistry Common Lab Assessment 1: Using the Density to Find the Amount of Sugar in a Carbonated Beverage Name: Elisa Johnson Lab Partner: Marleigh Bickel Block 1 Honors Chemistry Experiment Date: Thursday October 9, 2014 Report Date: Friday October 27, 2014 ABSTRACT: The purpose of this lab was to investigate the amount of sugar in carbonated beverages by using the density of solutions, and also to create a standard curve with the percent sugar in sugar solutions.

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Q. Provide the number of protons, neutrons, and electrons in Molybdenum-86. answer choices. $p = 86$, $n = 86$, neutrons = 42. $p = 42$, $n = 42$, $e = 44$. $p = 42$, $n = 44$, $e = 42$. $p = 42$, $n = 42$, $e = 42$.

~~Honors Chemistry—Atomic Structure Quiz—Quizizz~~

Group 1-the alkali metals and Group 2- the alkaline earth metals (nonmetals are anything to the left of the staircase) Which element is most similar to Li in chemical and physical properties: Na, Be, Si or Cl ? Na - it is in the same column and has the same number of valence electrons

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b) Construct an argument to support the claim that the proton (and not the neutron or electron) defines the element's identity. d) Construct an explanation that relates the relative abundance of isotopes of a particular element to the atomic mass of the element. SC2.

~~Unit 1—MHS Honors Chemistry~~

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