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Note the following examples: • " milli " means thousandth so a milliliter (symbol: mL) is one thousandth of a Liter and it takes one thousand mL to make one L. • " Mega " means million so " Megagram " (Mg) means one million grams NOT one millionth of a gram.

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An element has 120 Neutrons, 79 electrons, and an atomic number of 79. An element has 21 Neutrons, 20 Protons, and 18 electrons. A certain element exists as three different isotopes. 24.1% of all isotopes have a mass of 75.23 amu, 48.7% have a mass of 74.61 amu, and 27.2% have a mass of 75.20 amu. a.

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Chemquest 1: Number in chemistry Period 2 By: khoi and Duy. ChemQuest 12 Quantum Numbers Quantum numbers by Kelvin Wong. Chemquest 9 solutions to chemquest 9. Electron Configurations Worksheet More practice with pg 6 (the back of "Electron Energy Levels"). Thanks to Everett Community College for the worksheet.

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