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relate the number of moles of one
chemical to 0.0209 mol AgNO₃ Aspirin
production a. 13.5 kg C₉H₈O₄ b. 7.66 kg
C₄H₆O₃ c. 4.29 L HC₂H₃O₂ Ideal
stoichiometry calculations do not account

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Chapter 9 Review Stoichiometry Answer

Key Modern Chemistry

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Answer The study of the quantitative relationships between the amounts of reactants used and the amounts of products formed by a chemical reaction is called stoichiometry. _____ 2. Stoichiometry is based on the law of conservation of mass. _____ 3.

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Answer Key

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