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7 Assessment
Answers
Chapter 7
Assessment
Answers

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Chapter 1:

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Matter and
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Matter and
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*#3 Matter and
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Matter and

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1 - Matter and

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Theory What Is

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nature of matter
- Part-1 -
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**Changes of
States of Matter
Chemical and
Physical Changes**

States of Matter
: Solid Liquid
Gas *10 Amazing
Experiments with
Water*

Pure Substances
Page 9/47

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Matter And Mixtures!

(Classification
of Matter)

Physical and

Chemical Changes

Understanding
Atoms, elements,
and molecules
Part #1 (9min)

Basic Chemistry
Series - 1.2

What is matter?
- HD Properties

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of Matter And
Physical and
Chemical
Changes:

Chemistry for
Kids -

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Changes States
of Matter—

Solids, Liquids,
Gases \u0026
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Chemistry Change
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1 - Introduction
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The Stories of
Two Chemicals
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ozone formation,
chlorofluorocarbons, CFC's 1.2
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Chapter

Summaries -

**Chemistry Matter
and Change**

Chapter 1

Section 2 MATTER
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Matter And

Change Chapter

– Anything that

has MASS and
7 Assessment
VOLUME. ATOMS

Answers
and MOLECULES

the basic

building blocks

of matter.

ATOM-- the

smallest unit of

an element that

still maintains

the properties

that element.

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ELEMENT—a pure substance made up of only one type of atom.

Answers

**MATTER and
CHANGE Chapter 1
Section 1**

Chemistry:
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CHAPTER Class

STUDY GUIDE FOR
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Section 6.3

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CHAPTER 3
SOLUTIONS MANUAL

Matter—Propertie
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and Changes

Solutions Manual
Chemistry:

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3 35 Section 3.1

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Properties of

Matter pages

70–75 Problem-

Solving Lab 1.

Explain why the

flow of a

compressed gas

must be

controlled for

practical and

safe use.

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Chapter 1

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6. atomic mass
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Henry Moseley
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properties 13.
14 ...

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5 25 Electrons

in Atoms

Reviewing

Vocabulary Match

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Column B. Column

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The set of frequencies of the electromagnetic waves emitted by the atoms of an element 2. The minimum amount of energy that can be lost or gained by ...

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Matter - 2.1

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(examples) a
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depends on the
amount of matter
in a sample. a
property that
depends on the
type of matter
in a sample
not... mass,
volume, length,
weight.

chemistry matter

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Matter and

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chapter 2

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Transparency

Masters

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11 31 number of

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moles of unknown
substance O CO H

O C H CO H O C H

O H O C H O CO

use mole ratio 5

mol O 2 1 mol C

3 H 8 3 mol CO 2

1 mol C 3 H 8 4

mol H 2 O 1 mol

C 3 H 8 1 mol C

3 H 8 5 mol O 2

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MATH SKILLS

TRANSPARENCY

MASTER 16

A "chemical change" is a change that produces matter with a different composition than the original matter. Chemical Change A change in which one or more substances

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are converted
into different
substances. Heat
and light are
often evidence
of a chemical
change.

Properties of
Compounds Quite
different
properties than
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elements.

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