

Chapter 22 Hydrocarbon Compounds

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~~Chapter 22 Section 2 Hydrocarbons Part 2~~

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This hydrocarbon is called "2,3-pentadiene" because there are 2 (di) double bonds (ene) after 2 and 3 (2,3) of a 5 carbon line (penta). It is named the same way for triple bonds, but with an yne instead of ene. Next are isomers. Isomers are compounds with the same molecular formula but different molecular structures.

Chapter 22: Hydrocarbon Compounds - Chemistry by Anna

molecules of hydrocarbons, such as alkanes, are nonpolar molecules. Hydrocarbons. the simplest organic compounds contain only carbon and hydrogen. Alkane. a hydrocarbon in which there are only single covalent bonds. Straight-chain alkanes. contains any number of carbon atoms, one after the other, in a chain.

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Section 1: Hydrocarbons Hydrocarbons: Organic compound that contains only carbon and hydrogen. Alkane: Hydrocarbon containing only single covalent bonds. Straight-chain: Saturated hydrocarbon that contains any number of carbon atoms arranged one after the other in a chain. Homologous Series: Group of compounds in which there is a constant increment of change in molecular structure from one ...

chapter 22: hydrocarbon compounds

Made up of about 80% methane, 10% ethane, 4% propane, and 2% butane. The remaining 4% consists of nitrogen, helium, and hydrocarbons of higher molar mass. Propane and butane are separated from the other gases in natural gas by liquefaction. Complete combustion of a hydrocarbon produces a blue flame.

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Chapter 22 hydrocarbon compounds. STUDY. Flashcards. Learn. Write. Spell. Test. PLAY. Match. Gravity. Created by. Liv20cederstrom. Terms in this set (23) Alkanes. a hydrocarbon containing only single covalent bonds. Alkenes. Hydrocarbons with one or more carbon-carbon double bonds. alkyl group. a hydrocarbon substituent.

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Chemistry (12th Edition) Chapter 22 - Hydrocarbon ...

an organic compound that contains only carbon and hydrogen. alkane. a hydrocarbon containing only single covalent bonds; saturated hydrocarbons. straight-chain alkane. a saturated hydrocarbon that contains any number of carbon atoms arranged one after another in a chain. homologous series.

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1. Find the root word (ending in -ane) in the hydrocarbon name. Then write the longest carbon chain to create the parent structure. 2. Number the carbons on this parent carbon chain. 3. Identify the substituent groups. Attach the substituents to the numbered parent chain at the proper positions. 4. Add hydrogens as needed.

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Chapter 22 "Hydrocarbon Compounds". Use these activities to learn the vocabulary and major concepts presented in the chapter. molecules in which the atoms are joined in the same order, but the arrangements of the atoms in space are different.

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Chapter 22 Hydrocarbon Compounds 237. SECTION 22.1 HYDROCARBONS (pages 693–701) This section describes the bonding in hydrocarbons and distinguishes straight-chain from branched-chain alkanes. It also provides rules for naming branch-chained alkanes. Organic Chemistry and Hydrocarbons (pages 693–694)

SECTION 22.1 HYDROCARBONS (pages 693–701)

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Hydrocarbons are chemical compounds that contain the elements carbon and hydrogen only. They are compounds that are obtained from the fossil fuel crude oil by a process called fractional...

Hydrocarbons - Hydrocarbons - National 4 Chemistry ...

hydrocarbons results in a diverse group of organic compounds. CHAPTER 22 SSubstituted Hydrocarbons and Their Reactionsubstituted Hydrocarbons and Their Reactions The larva of the Cerura vinula moth squirts formic acid when threatened. Formic acid is a substituted hydrocarbon—a type of compound that forms when one or more hydrogens

SSubstituted Hydrocarbons and Their Reactionsubstituted ...

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This unit looks at aliphatic and aromatic hydrocarbons, including alkyl substituted hydrocarbons and structural and geometric isomers. There is also a brief introduction to functional groups. Chapter 22 - Hydrocarbon Compounds

Chapters 22 & 23 - Organic Chemistry - Mrs. Gingras ...

Hydrocarbon compounds Section 22.1: Hydrocarbons What is the relationship between carbon and organic chemistry? 200 years ago scientists thought that only living organisms could make carbon compounds known in living cells Organic means related to or made by living things In 1828, A German Chemist, Friedrich Wohler

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