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Chapter 21 □ Nuclear Chemistry: Part 1 of 9

Nuclear Chemistry: Crash Course

Chemistry #38 Chapter 21: Nuclear

Chemistry (Chem in 15 minutes or less)

CHM 152 / Chapter 21 / 1 / Radiation and

Radioactive Isotopes ~~Chapter 21 □ Nuclear~~

~~Chemistry: Part 3 of 9~~

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Chapter 21 □ Nuclear Chemistry: Part 2 of 9
Chapter 21 □ Nuclear Chemistry: Part 6 of 9
Chapter 21 □ Nuclear Chemistry: Part 9 of 9 ~~Carbon-13 NMR Spectroscopy~~

Chapter 21 □ Nuclear Chemistry: Part 4 of 9
MY GCSE RESULTS 2018 *very emotional*
21 GCSE Physics Equations Song
~~Nuclear Half Life: Calculations~~

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Nuclear Chemistry Part 2 - Fusion and
Fission: Crash Course Chemistry #39

Einstein's Proof of $E=mc^2$ How to

Calculate the Mass Defect and Binding

Energy Chapter 14 □ Chemical Kinetics:

Part 1 of 17 ~~#Q value of a Nuclear~~

~~Reaction, Unit 8, Atomic~~

~~Nucleus, Chp 4, Nuclear~~

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~~Reactions, Class 12, Physics~~ Half Life
Chemistry Problems - Nuclear Radioactive
Decay Calculations Practice Examples

Nuclear Chemistry, 'Alpha Decay' by Dr.
Anju Agrawal, M.Sc. Pt-III, Chemistry
~~Chapter 21 □ Nuclear Chemistry: Part 5 of~~
9 Chapter 21 □ Nuclear Chemistry: Part 7
of 9 The Corporation - Feature Film

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Chapter 21 - Nuclear Chemistry - Section
2 ALL OF CIE IGCSE PHYSICS 9-1 /
A*-U (2021) | IGCSE Physics Revision |
Science with Hazel Chapter 21 Nuclear
Chemistry: Part 8 of 9 Radiation Exposure
FSC Physics Part 2, Chapter 21, Nuclear
Physics

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CHAPTER 21 REVIEW Nuclear
Chemistry. Modern Chemistry 175
Nuclearchemistry CHAPTER 21
REVIEW Nuclear Chemistry SECTION 4
SHORT ANSWER Answer the following
questions in the space provided. 1. Match
each of the following statements with the
process(es) to which they apply, using one

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of the choices below: (1) fission only (3)
both fission and fusion

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Answers

Chapter 21 Nuclear Chemistry Review
Answers The n:p ratio for Cr-53 is 29 24

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$29\ 24 = 1.21$; for Mn-51, it is $26\ 25\ 26\ 25 = 1.04$; for Fe-59, it is $33\ 26\ 33\ 26 = 1.27$.
Positron decay occurs when the

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nucleons. protons and neutrons. nuclide.

An atom identified by the number of
protons

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Nuclear Chemistry Nuclear

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Transformations □ Rutherford in 1919 performed the first nuclear transformation.

□ The transmutations are sometimes represented by listing in order, the target nucleus, the bombarding particle, the ejecting particle and the product nucleus. □

The above equation becomes: ${}_{14}^{27}\text{Al} + {}_2^4\text{He} \rightarrow {}_{17}^{31}\text{P} + {}_1^1\text{H}$

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Study Guide Chapter 5-21 Answer Key

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21.7: Nuclear Fission; 21.8: Nuclear Fusion; 21.9: Biological Effects of Radiation; 21.E: Exercises; 21.S: Nuclear Chemistry (Summary) □ 22

21: Nuclear Chemistry - Chemistry
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Chapter 21 Review Nuclear Chemistry Answers

A nuclear fuel. A fissionable isotope must be present in large enough quantities to sustain a controlled chain reaction. The radioactive isotope is contained in tubes called fuel rods. A moderator. A

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moderator slows neutrons produced by nuclear reactions so that they can be absorbed by the fuel and cause additional nuclear reactions. A coolant.

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