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Activated complex (p. 532) A short-lived, unstable arrangement of atoms that may break apart and re-form the reactants or may form products; also sometimes referred to as the transition state.

Activation energy (p. 533) The minimum amount of energy required by reacting particles in order to form the activated complex and lead to a reaction ...

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Calorimeter. a device used to measure the heat absorbed or released in a chemical or physical change. Enthalpy change. The amount of energy absorbed or lost by a system during a process at constant pressure. Heat. The energy transferred between samples of matter because of a difference in their temperature . Heat of combustion. Energy released as ...

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