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Chemistry -

Basic

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vanilla, onion and clove, can

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Acids Bases

be used as
olfactory
indicators on
the basis of

your

observations?

Let us do some
more activities
to understand
the chemical
properties of
acids and bases.

2.1.2 How do

Acids and Bases

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React with

Metals? n Rinse

both cloth

strips with

water and again

check their

odour.

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As this chapter

19 acids bases

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it ends

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0:01:31

Introduction

0:05:45

difference

between Acids &

Bases 0:12:06

Aqueous solution

of Acids & bases

0:18:06

Naturally

occur...

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~~SALTS FULL~~

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Acid + base ?

salt + water +

heat. H_2SO_4

+ MgO ? $MgSO_4$ + H_2O

$2HCl$ + $Mg(OH)_2$

? $MgCl_2$ + $2H_2O$.

2. Reaction of

non-metal oxides

with bases. Non-

metal oxides are

acidic in nature

Base + Non-metal

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oxide ? salt +
water + heat.

$2\text{NaOH} + \text{CO}_2 \rightarrow \text{Na}_2\text{CO}_3 + \text{H}_2\text{O}$. To

know more about
Properties of
Acids and Bases,
visit here.

Water Acids and
bases in water

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1. Common Salt

(Sodium

Chloride):

Sodium chloride

(NaCl) is also

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known as Common
or Table Salt.

It is formed
after the
reaction between
sodium ... 2.

Bleaching Powder
(CaOCl_2):

Bleaching powder
is also known as
chloride of
lime. It is a
...

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salts (handouts)

Chapter

19...Acids, Bases,

and Salts ACID-

BASE THEORIES

Acids and bases

are all around

us and part of

our everyday

life (ex. bodily

functions,

vinegar,

carbonated

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drinks, citrus
fruits, car
batteries,
antacids,
cleaning
products, etc.).
Chapter 19 Acids
Bases And Salts
Answer Key Salts
. When acid and
base neutralize,
salts are
formed.

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MCQs Questions
with Answers.

Choose the
correct answer:
Question 1. An
oxide is acidic
and has a
pungent odour.

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~~5 Acids, Bases~~

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5: In today's

Class 7 Science

Sprint session

in Chapter 5, we

discuss acids

bases and salts

class 7. The

aspirants...

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Chapter 19

"Acids, Bases,
and Salts". Use
these activities
to study the
vocabulary and
major concepts
presented in
this chapter.
compounds

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derived from the reaction of a strong base with a weak acid or a strong acid with a weak base or a solution of a weak base and one of its salts.

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