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2? A 2+ B 3+ C 2? D 3? 4. Fe. 2. O. 3 + x. CO

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. y. $\text{Fe} + 3\text{CO}$. 2. This equation will be balanced when. A . $x = 1$ and . $y = 2$ B . $x = 2$ and . $y = 2$ C . $x = 3$ and . $y = 2$ D . $x = 2$ and . $y = 3$. 5. An acidic solution contains A only hydrogen ions B only hydroxide ions C more hydrogen ions than hydroxide ions D more hydroxide ions than hydrogen ions. 6.

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2 (g) ? $3\text{CO}_2(\text{g}) + 4\text{H}_2\text{O}(\text{l})$ Name the type of enthalpy change which the student measured.

No marks should be awarded for 'burning' since the word 'burned' appears in the text.

(o) Unless the question is clearly about a non-chemistry issue, eg costs in industrial chemical process, a non-chemical answer gains no marks.

2014 Chemistry National 5 Finalised Marking Instructions

June 19, 2017 in AHSEC, AHSEC 12 - Chemistry, ahsec notes, AHSEC Science Question Papers

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Question 3. Write the IUPAC name of the compound [1] Answer: 3-aminebutan-1-al.

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Question 5.

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2 by electricity / (electric) current / (flow of) electrons Mark independently b A = chlorine / Cl₂ Ignore Cl 3 B = hydrogen / H₂ Ignore H C = sodium hydroxide / NaOH Ignore references to sodium chloride If both name and formula given, both must be correct, but ignore Cl and H Award 1 mark for chlorine and hydrogen the wrong way round

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